

SEMESTER I

ENVIRONMENTAL STUDIES

1. Course Description

Program: BSC
Course Code: U24/EVS/AECC/101
Course Type: AECC
No. of credits: 2

Max. Hours: 30
Hours per week: 2
Max. Marks: 50

2. Course Objectives:

- To Understand the principles of ecology and environmental issues
- To acquire the skills needed and develop a sense of responsibility to actively participate in its protection and improvement

3. Course Outcomes:

On completion of the course the student will be able to:

CO 1: Gain knowledge and develop in-depth understanding of the basics of ecological principles, conservation of biodiversity, renewable energy resources and water conservation

CO 2: Enhanced analytical capability to undertake and participate in finding solutions for various environmental issues and concerns of national and global importance to achieve environmental protection and sustainable development

4. COURSE CONTENT

UNIT - I: Ecosystem, Biodiversity & Natural Resources (15 hrs.)

1. Definition, Scope & Importance of Environmental Studies
2. Structure of Ecosystem – Abiotic & Biotic Components, Ecological Pyramids
3. Definition of Biodiversity, Genetic, Species & Ecosystem Diversity, IUCN Red list, Hotspots of Biodiversity, Threats and Conservation of Biodiversity (*In situ & Ex situ*)
4. Renewable Energy Resources – Solar, Wind and Biomass
5. Water Conservation, Water Footprint, Rain Water Harvesting
6. Environmental Ethics

UNIT – II: Environmental Pollution, Global Issues & Legislation (15 hrs.)

1. Causes, Effects and Control Measures of Air and Water Pollution
2. Solid and Plastic Waste Management, Zero Waste Management
3. Global Warming & Ozone Layer Depletion, Carbon Footprint
4. Environmental Laws and Acts-Wildlife Protection Act, Forest Act, Air Act, Water Act
5. People's Participation in Environmental Protection- Silent Valley, Bishnois of Rajasthan
6. Disaster Management-Flood, Earthquake and Cyclones
7. Environmental Management
8. Role of Information Technology in Environmental Protection and Health

Field visit:

1. Visit to Solar Plant in your Locality/City
2. A Glimpse of Biodiversity in Hyderabad/ Visit to National Parks and a Walk-Through Campus
3. Visit to a Nearby Lake

5. REFERENCES:

Books:

- Text book of Environmental Studies for undergraduate courses (second edition) by Erach Bharucha
- Environmental Studies by Dr. J.P. Sharma
- Perspectives in Environmental Studies – Anubha Kaushik & C.P. Kaushik
- A text book of Environmental Studies by Dr. D. K. Asthana and Dr. Meera Asthana
- Environmental Science by Dr. Syeda Azeem Unnisa

Magazines:

- **Terra Green (a monthly digital magazine on environmental issues)**
- Down to Earth, Centre for Science &
- Environment Survey of the Environment published by The Hindu

E-Resources:

- <https://www.cseindia.org/>
- <https://www.ugc.gov.in/oldpdf/modelcurriculum/env.pdf>

6. Syllabus Focus

a) Relevance to Local, Regional, National and Global Development Needs

Local /Regional/ National /Global Development Needs	Relevance
Local needs	<p>Develop a critical understanding of Environmental issues and concerns. Inculcate the environmental ethics and work for sustainable future</p> <p>Utilise the potential application of Methods of Solid Waste Management in the Waste management concerns</p> <p>Involve in community development through extension and organising programs.</p>
Regional needs	<p>Creates awareness on pollution and threats to biodiversity in the Ecosystem</p>
National needs	<p>Have an over view of mitigation measures of disaster management. Explain major conservation strategies taken in India. Apply the Knowledge of role of information technology in protection of the environment.</p>
Global needs	<p>Environmental studies is globally relevant to monitor environmental issues and for the sustainable development. It deals with issues and challenges of environment management in the changing climate scenario.</p>

b) Components on Skill Development/Entrepreneurship Development/Employability

SD/ED/EMP	Syllabus Content	Description of Activity
Skill Development, Entrepreneurship Development, Employability	Unit II Solid Waste Management	Demonstration of Composting, Vermicomposting and the preparation of Bio-Enzymes. Awareness on scope of green entrepreneurship and employability related to Solid waste management

7. Pedagogy

S. No	Student Centric Methods Adopted	Type / Description of Activity
1.	Experiential learning	Field trips, Documentary watching, Demonstrations, Student volunteering days, Plantation drives, Clean up drives
2.	Participative Learning	Seminars, Workshops, Guest lectures, Group Discussion, Library reference, Presentations and Competitions, Demonstrations by students
3.	Problem Solving	Case Studies, Projects

8. Course Assessment Plan

a) Weightage of Marks in Internal Assessments and End Semester Examination

CO	Internal Assessments IA -40%	End Semester Examination-60%
CO1	Field Visit report/Case Study/ Poster making/ Presentations/Eco Friendly product making/Model making	Written Exam
CO2		

7 DEPARTMENT OF ENVIRONMENTAL STUDIES
CHAIR PERSON B.O.E in
Environmental Science
University College of Science
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ST. FRANCIS COLLEGE FOR WOMEN, HYDERABAD
Head
Dept. of Environmental Studies
St. Francis College for Women
(Autonomous)
Begumpet, Hyderabad-16

b) Model Question Paper- End Semester Exam

ENVIRONMENTAL STUDIES

Course Code: U24/EVS/AECC/101

Time: 1 Hour

Max. Marks: 30

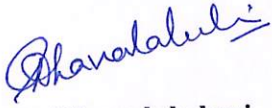


Answer any five of the following:

5X6=30

1. Define environmental studies and mention the importance of environmental studies.
2. "Biomass is an important source of energy", Discuss.
3. Explain the rooftop rainwater harvesting system with the help of a diagram.
4. Identify the reasons for water pollution in your region and suggest measures to reduce the water pollution.
5. Discover the initiatives taken by swachh cities in solid waste management.
6. Comment on "Silent Valley- A people's movement that saved a forest."
7. List out the changes you would make in your lifestyle to reduce your carbon footprint.

c) Question Paper Blueprint

Modules	Hours Allotted in the Syllabus	COs Addressed	Section A (No. of Questions)	Total Marks
I	15	CO 1	3	6
II	15	CO 2	4	6

Prepared by	Checked & Verified by	Approved by
 G. Dhanalakshmi Head, Dept. of Environmental Studies	 G. Dhanalakshmi, Head, Dept. of Environmental Studies	 Dr. Uma Joseph Principal

SEMESTER II

ENVIRONMENTAL STUDIES

6. Course Description

Program: BA, BMS & BCOM
Course Code: U24/EVS/AECC/201
Course Type: AECC
No. of credits: 2

Max. Hours: 30
Hours per week: 2
Max. Marks: 50

7. Course Objectives:

- To Understand the principles of ecology and environmental issues
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9. COURSE CONTENT

UNIT - I: Ecosystem, Biodiversity & Natural Resources (15 hrs.)

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CO2		

b) Model Question Paper- End Semester Exam

ENVIRONMENTAL STUDIES

Course Code: U24/EVS/AECC/201

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Max. Marks: 30


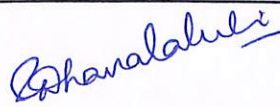

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II	15	CO 2	4	6

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 G. Dhanalakshmi Head, Dept. of Environmental Studies	 G. Dhanalakshmi, Head, Dept. of Environmental Studies	 Dr. Uma Joseph Principal

SEMESTER - II

ANIMAL DIVERSITY - VERTEBRATES

1. Course Description

Programme: B.Sc

Course Code: U24/ ZOO/ DSC/201

Course Type: DSC - II

No. of credits: 4

Max. Hours: 60

Hours per week: 4

Max. Marks: 100

2. Course Objectives

- Identify the various vertebrate classes and cites the examples of individual species that belong to them
- Describes the diversity and evolutionary relationship of vertebrates by integrating knowledge in comparative anatomy and physiology
- Understands the geographical distribution of common vertebrates and basic structural & functional aspects of Animal diversity

3. Course Outcomes

On completion of the course the student will be able to:

CO1: To understand and appreciate the study of significance of animal biodiversity.

CO2: To remember general characteristics of vertebrates.

CO3: To analyze the various adaptations in vertebrates and their basic differences among Amphibians, Reptiles, Birds and Mammals.

CO4: To apply the gained knowledge in comprehending the structural organization in vertebrates.



4. Course Content**Module I: Urochordata, Cephalochordata and Cyclostomata 15 HRS**

- 1.1 Salient features of Urochordata
- 1.2 Retrogressive metamorphosis and its significance in Urochordata
- 1.3 Salient features and affinities of Cephalochordata; General characters of Cyclostomata
- 1.4 Comparison of the Petromyzon and Myxine
- 1.5 General characters and classification of Chordata upto orders with examples.

Module II: Pisces & Amphibia 15 HRS

- 2.1 General characters of Fishes; Classification of fishes up to order level with examples
- 2.2 Scoliodon – Respiratory, Circulatory and Nervous system.
- 2.3 Types of Scales and types of Fins
- 2.4 General characters and Classification of Amphibians up to orders with examples.
- 2.5 Rana tigrina - Respiratory, Circulatory and Nervous system.
- 2.6 Parental care in amphibians; Neoteny and Paedogenesis.

Module III: Reptilia & Aves 15 HRS

- 3.1 General characters of Reptilia; Classification of Reptilia up to orders with examples
- 3.2 Calotes – Respiratory system, Circulatory and Nervous system
- 3.3 Temporal fosse in reptiles and its evolutionary importance;
- 3.4 General characters of Aves; Classification of Aves up to orders with examples.
- 3.5 Columba livia - Digestive system, Circulatory systems, Respiratory system and Nervous system.
- 3.6 Migration in Birds; Flight adaptation in Birds


Module IV: Mammalia 15 HRS


- 4.1 General characters of Mammalia
- 4.2 Classification of Mammalia up to orders with examples
- 4.3 Rabbit – Digestive system, Respiratory, Circulatory and Nervous system.
- 4.4 Dentition in mammals.
- 4.5 Aquatic adaptations in Mammals.
- 4.6 Flying adaptations in Mammals



5. References:

1. E.L.Jordan and P.S. Verma 'Chordate Zoology' -. S. Chand Publications.
2. Mohan P.Arora. 'Chordata – I, Himalaya Publishing House Pvt.Ltd.
3. Marshal, Parker and Haswell 'Text book of Vertebrates'. ELBS and McMillan, England.
4. Alfred Sherwood Romer. Thomas S. Pearson 'The Vertebrate Body, Sixth edition, CBS college Publishing, Saunders College Publishing
5. J.W. Young, The Life of Vertebrates, 3rd ed, Oxford University press.
6. Harvey Pough F, Christine M. Janis, B. Heiser, Vertebrate Life, Pearson, 6th ed, Pearson Education Inc.2002.



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Osmania University, Hyd-07.

6.Syllabus Focus

a)Relevance to Local , Regional , National and Global Development Needs

Local /Regional/National /Global Development Needs	Relevance
National needs	Identifying priority areas for biodiversity is essential for directing conservation resources. Fundamentally, we must know where individual species live, which ones are vulnerable, where human actions threaten them, and their levels of protection.
Global development needs	To map global priority areas using the latest data on vertebrates. As conservation knowledge and threats change, it is essential it is reevaluate on global priorities.



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b)Components on Skill Development/Entrepreneurship Development/Employability


SD/ED/EMP	Syllabus Content	Description of Activity
Skill Development	Module II	Dissection techniques and structural organization in vertebrates
Entrepreneurship Development	Module III & IV	Adaptation and Conservation of Vertebrates

7. Pedagogy

S. No	Student Centric Methods Adopted	Type / Description of Activity
1.	Experiential Learning	Field trips / Visit to Nehru Zoological Parks, Wild life sanctuaries.
2.	Participative learning	Presentation / Ppt presentation on general topics.
3.	Problem Solving learning	Awareness programmes / local activities of Conservation of Animals.

8. Course Assessment Plan**a)Weightage of Marks in Continuous Internal Assessments and End Semester Examination**

CO	Continuous Internal Assessments CIA -40%	End Semester Examination-60%
CO1	CIA-1 Written Test	Written Exam
CO2	CIA-1 Written Test	
CO3	CIA - 2 - Assignment	
CO4	CIA - 2 - Objective test	


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b) Model Question Paper- End Semester Exam

ANIMAL DIVERSITY – VERTEBRATES
MODEL PAPER - THEORYCOURSE: DSC- II
SUBJECT CODE: U24/ZOO/DSC/201MAX MARKS: 60
TIME: 2HRS

ILLUSTRATE WITH DIAGRAMS WHEREVER NECESSARY

SECTION-A

I. Answer the following: -

4 X 10 = 40 M

1. Explain Retrogressive metamorphism and its significance in Urochordata
OR
2. Give characters of cyclostomata and discuss their phylogenetic status.
3. Describe the structure of Heart and course of blood circulation in Scoliodon.
OR
4. Discuss parental care in Amphibia with suitable examples.
5. Give an account of Heart , blood and course of circulation of Calotes
OR
6. Describe various modes of flight in birds.
7. Describe the structure and working of mammalian Heart. With a note on its automaticity
OR
8. Explain dentition in Mammals and add a note on development of a tooth.

SECTION- B

II. Answer any four questions

4 X 5 = 20 M

9. Salient features of Urochordata
10. Types of Scales
11. Neoteny
12. Temporal fossa in reptiles
13. Migration in Birds
14. Flying adaptations in Mammals

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Department of Zoology, UCS,
Osmania University, Hyd-07.

ANIMAL DIVERSITY - VERTEBRATES - MODEL PAPER THEORY

SECTION A - INTERNAL CHOICE				4 Q X 10 M = 40 M
Question Number	Question	Question	CO	BTL (Bloom's Taxonomy Level)
1	Module 1	Explain Retrogressive metamorphism and its significance in Urochordata	CO 1	2
2	Module 1	Give characters of cyclostomata and discuss their phylogenetic status	CO 1	2
3	Module 2	Describe the structure of Heart and course of blood circulation in Scoliodon	CO 2	2
4	Module 2	Discuss parental care in Amphibia with suitable examples	CO 2	1
5	Module 3	Give an account of Heart , blood and course of circulation of Calotes	CO 3	2
6	Module 3	Describe various modes of flight in birds.	CO 3	1
7	Module 4	Describe the structure and working of mammalian Heart. With a note on its automaticity	CO 4	4
8	Module 4	Explain dentition in Mammals and add a note on development of a tooth.	CO 4	4
SECTION B - ANSWER ANY 4 OUT OF 6				4 Q X5 M = 20 M
9	Module 1	Salient features of Urochordata	CO 1	1
10	Module 2	Types of Scales	CO 2	1
11	Module 3	Neoteny	CO 3	3
12	Module 4	Temporal fosse in reptiles	CO 4	2
13	Module 3	Migration in Birds	CO 3	4
14	Module 4	Flying adaptations in Mammals	CO 4	4

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PROFESSOR & HEAD
Department of Zoology, UCS
Omania University, Hyd-07.

ANIMAL DIVERSITY- VERTEBRATES PRACTICALS

Programme: B.Sc.

Type of course : DSC - II

CourseCode: U24/ZOO/DSC/201/P

Max. Marks : 50

Hours per week : 2

No. of credits : 1

Course Objective:

To understand the basic structural & functional aspects of Animal diversity – Vertebrates

Course Outcome:

To identify and remember the various vertebrate specimens under study.

To evaluate the techniques learnt during dissections.

Study of museum slides / specimens / models (Classification of animals up to orders)

1. Protochordata: *Amphioxus*, *Amphioxus* T.S. through pharynx
2. Cyclostomata: *Petromyzon*, *Myxine*, *Ammocoetes* larva
3. Pisces: *Sphyrna Pristis*, *Torpedo*, *Pleuronectes*, *Hippocampus*, *Exocoetus*, *Echeneis*, *Labeo*, *Catla*, *Clarius*, *Anguilla*, *Protopterus*
4. Amphibia: *Ichthyophis*, *Amblystoma*, *Siren*, *Hyla*, *Rachophous*, *Bufo*, *Rana*, Axolotl larva
5. Reptilia : *Draco*, *Chameleon*, *Gecko*, *Uromastyx*, *Vipera russelli*, *Naja*, *Bungarus*, *Enhydrina*, *Typhlops*, *Testudo*, *Trionyx*, *Crocodilus*, *Ptyas*.
6. Aves: *Archaeopteryx*, *Passer*, *Psittacula*, *Bubo*, *Alcedo*, *Columba*, *Corvus*, *Pavo*; Collection and study of different types of feathers: Quill, Contour, Filoplume, Down
7. Mammalia: *Ornithorhynchus*, *Tachyglossus*, *Pteropus*, *Funambulus*, *Manis*, *Loris*, Hedgehog

Osteology:

1. Rabbit – Axial skeleton system (bones of Skull and Vertebral Column)
2. Pigeon and Rabbit – Appendicular skeleton system (bones of limbs and girdles)

Dissections of *Labeo/Tilapia*: Brain, Weberian ossicles & V, VII, IX, X cranial nerves

An “Animal album” containing photographs, cut outs, with appropriate write up about the above mentioned taxa. Different taxa/ topics may be given to different sets of students for this purpose.

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DEPARTMENT OF ZOOLOGY, ST. FRANCIS COLLEGE FOR WOMEN, HYDERABAD

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Board of Studies in Zoology,
Osmania University, Hyd-07.

PROFESSOR & HEAD
Department of Zoology, U
Osmania University, Hyd-07

**ANIMAL DIVERSITY- VERTEBRATES
PRACTICALS – MODEL PAPER**

Subject Code: U24/ZOO/DSC/201/P
No. of Credits:1

Max. Marks – 50
Time – 2 hour

I. Major Dissection**14 M**

1. Dissect and display the Vth and VIIth Cranial Nerves of Labeo (OR)
2. Dissect and display the IXth and Xth Cranial Nerves of Labeo

II. Minor Dissection**6 M**




3. Dissect and display the Brain of Labeo (OR)
4. Dissect and display the Weberian Ossicle of Labeo


III. Identify, classify, draw, label & write points of identification for the following spots (A-G).

7x2 =14 M

- A. Protochordata: Amphioxus, Amphioxus T.S. through pharynx
B. Cyclostomata: Petromyzon, Myxine, Ammocoetus larva
C. Pisces: Sphyrna, Pristis, Torpedo, Pleuronectes, Hippocampus, Exocoetus, Echieneis, Labeo, Catla, Clarius, Auguilla, Protopterus
D. Amphibia: Ichthyophis, Amblystoma, Siren, Hyla, Rachophous, Bufo, Rana, Axolotallarva
E. Reptilia :Draco, Chamaeleon, Gecko, Uromastix, Vipera russelli, Naja, Bungarus, Enhydrina, Typhlops, Testudo, Trionyx, Crocodilus, Ptyas.
F. Aves: Archaeopteryx, Passer, Psittacula, Bubo, Alcedo, Columba, Corvus, Pavo; Collection and study of different types of feathers: Quill, Contour, Filoplume, Down
G. Mammalia: Ornithorhynchus, Tachyglossus, Pteropus, Funambulus, Manis, Loris, Hedgehog

III. OSTEOLOGY- IDENTIFICATION OF BONES (H-J)**2X2 = 4 M****IV. ANIMAL ALBUM****5 M****V. RECORD****5 M**

Prepared by	Checked & Verified by	Approved by
 Name and Signature of the teaching faculty	 Name and Signature of HoD DR. JYOTHI RANI	 Name and Signature of Principal


PROFESSOR & HEAD
Board of Studies in Zoology, Department of Zoology, UCS,
Osmania University, Hyd-07. Osmania University, Hyd-07.

**St. FRANCIS COLLEGE FOR WOMEN, BEGUMPET HYDERABAD-
500016 (An Autonomous College Affiliated to Osmania University)**

DEPARTMENT OF CHEMISTRY

**DISCIPLINE SPECIFIC CORE I
CHEMISTRY PAPER I
B.Sc. I - SEMESTER- II 60 Hrs**

Module 1: Inorganic Chemistry

p-block elements
Interhalogen Compounds
Chemistry of Zero group elements
Boranes and Carboranes
Metal carbonyls

Module 2: Physical Chemistry

Electrochemistry

Module 3: Organic Chemistry

Aromatic Hydrocarbons
Halogen compounds
Alcohols, Phenols, Ethers

Module 4: General Chemistry

Solutions
Symmetry of molecules
Stereochemistry of Carbon Compounds

CHEMISTRY – II

Course Description

Programme: B.Sc. Max. Hours: 60 Hrs


Course Code: U24/CHE/DSC/201 Hours per week: 4 Hrs

Course Type: DSC-2 Max. Marks: 100

No. of credits: 4

Course Objectives

- To study about the elements of p block and the properties of their compounds.
- To understand the behavior of electrolytes in solution and to know the applications of electrode


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process.

- To learn the methods of preparation and reactivity of hydrocarbons with mechanisms and to classify and identify organic molecules by their functional groups.

• To learn the structures of basic organic molecules, the types of reactions they undergo and their stereochemistry and to foster acquisition of knowledge on the concepts of solutions.

Course Outcomes

On completion of the course the student will be able to:

CO1: Acquire knowledge on p-block elements.

CO2: Understand the theory of electrical conductance, transformation of chemical energy into electrical energy in Galvanic cells.

CO3: Interpret the concept of aromaticity and familiarize with the various types of aliphatic and aromatic reactions. Recognize functional groups in organic molecules and predict their reactivity through mechanisms.

CO4: Application of the behaviour of solutions and acquire a fundamental understanding of the relationships between molecular structure and reaction mechanisms.

Course Content

MODULE I: Inorganic Chemistry

(15 Hrs)

p-block elements

(10 Hrs)

General Characteristics of p block elements.

Group – 13: Synthesis and structure of diborane and higher Boranes (B₂H₆ and B₃H₉). Preparation and structure of boron-nitrogen compounds (B₃NH₆ and BN), Lewis acid nature of the BX₃.

Group – 14: Classification (ionic, covalent, interstitial) and industrial applications of Carbides. Preparation, classification (straight chain, cyclic and cross-linked) and applications of silicones, Preparation and applications of graphitic compounds.

Group – 15: Preparation, structure and reactions of hydrazine, hydroxylamine, Phosphazenes

Group – 16: Classifications of oxides based on (i) Chemical behavior and (ii) Oxygen content. Normal: acid, basic, amphoteric and neutral, Mixed oxides, Sub oxides, Peroxides, Super oxides.

Oxyacids of N, P, S and Cl – structure, acidic nature and redox properties

Interhalogen Compounds

Classification- general preparation- structures of AB, AB₂, AB₃ and AB₄ type and reactivity. Poly halides- definition and structure of ICl₃, ICl₄ and I₂. Comparison of Pseudo halogens with halogens.

Chemistry of Zero group elements


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General preparation, structure, bonding and reactivity of Xenon compounds – Oxides, Halides. Clathrate compounds.

Boranes and Carboranes

(2 Hrs)

Definition of clusters. Structures of boranes and carboranes- Wade's rules, closo, nido, arachno Boranes and carboranes

Metal carbonyls

(3 Hrs)

Preparation and properties of Ni(CO)_4 . Structural features of Ni(CO)_4 , Fe(CO)_5 , $\text{Fe}_2(\text{CO})_9$, $\text{Fe}_3(\text{CO})_{12}$ and Cr(CO)_6 . -18 valence electron rule.

MODULE II: Physical Chemistry

(15 Hrs)

Electrochemistry

Electrical transport – conduction in metals & in electrolyte solutions, specific conductance & equivalent conductance and measurement of equivalent conductance, variation of specific and equivalent conductance with dilution. Migration of ions and Kohlrausch's law. Arrhenius theory of electrolytic dissociation and its limitation, weak and strong electrolytes, Ostwald's dilution law, its uses and limitations. Debye-Huckel-Onsager's equation for strong electrolytes (elementary treatment only). Transport number, definition and determination by Hittorf method for attackable and non-attackable electrodes. Applications of conductivity measurements. Determination of degree of dissociation, determination of K_a of acids, determination of solubility product of sparingly soluble salt, conductometric titrations. Electrolytic and Galvanic cell – reversible and irreversible cells, conventional representation of electrochemical cell. EMF of a cell and its measurement. Computation of EMF. Types of reversible electrodes gas electrode, metal - metal ion, metal - insoluble salt and redox electrode. Electrode reactions, Nernst equation, cell EMF and single electrode potential, standard Hydrogen electrode – reference electrodes (calomel electrode) – standard electrode potential, sign conventions, electrochemical series and its significance. Calculation of thermodynamic quantities of cell reaction – ΔG , ΔH and K . Determination of pH using Hydrogen electrode, Glass electrode, quinhydrone electrode, solubility product of AgCl . Potentiometric titrations.

MODULE III: Organic Chemistry

(15 Hrs)

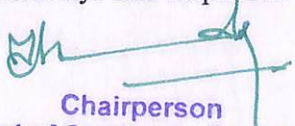
Aromatic Hydrocarbons

(6 Hrs)

Concept of aromaticity – definition, Huckel's rule – application to Benzenoids and Non – Benzenoids (cyclopropenyl cation, cyclopentadienyl anion and tropylium cation).

Preparations: From acetylene, phenols, benzene carboxylic acids and sulphonic acids

Reactions - General mechanism of electrophilic substitution, mechanism of nitration, sulphonation, and halogenation, Friedel Craft's alkylation and acylation. Orientation of aromatic substitution - Definition of ortho, para, and meta directing groups. Ring activating and deactivating groups with examples. Orientation – (i) activating groups: Amino, methoxy and alkyl groups. (ii) Deactivating groups - carboxyl, nitro, nitrile, carbonyl and sulphonic acid & halo groups.


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Halogen compounds

(4 Hrs)

Nomenclature and classification: alkyl (primary, secondary, tertiary), aryl, aralkyl, allyl, vinyl, benzyl. Chemical reactivity - reduction, formation of RMgX, Nucleophilic substitution reactions – classification into S₁ and S₂. Mechanism and energy profile diagrams of S₁ and S₂ reactions. Stereochemistry of S₂ (Walden Inversion) 2-Bromobutane, S₁ (Racemisation) 1-Bromo-1-phenylpropane explanation of both by taking the example of optically active alkyl halide. Structure and reactivity – Ease of hydrolysis - comparison of alkyl, vinyl, allyl, aryl, and benzyl halides.

Alcohols, Phenols, Ethers

(5 Hrs)

Alcohols: Preparation of 1, 2 and 3 alcohols using Grignard reagent, Ester hydrolysis, Reduction of aldehydes, ketones, carboxylic acid and esters. Reactions: With sodium, HX (Lucas test), esterification (mechanism), oxidation (with PCC, alk. KMnO₄, acidic dichromate, conc. HNO₃). Oppenauer oxidation

Diols: oxidation of diols, Pinacol – Pinacolone rearrangement.

Phenols: Preparation: Cumene hydroperoxide method, from diazonium salts. Reactions: Electrophilic substitution: Nitration, Halogenation and sulphonation. Reimer-Tiemann Reaction (with mechanism), Gattermann Aldehyde Reaction, Houben-Hoesch Condensation, Schotten – Baumann Reaction, Azo coupling reactions

Ethers (aliphatic and aromatic): Preparation: Williamson synthesis, Reaction: Cleavage of ethers with HI.

MODULE IV: General Chemistry

(15 Hrs)

Solutions

(6 Hrs)

Liquid- liquid mixtures – ideal liquid mixtures, Raoult's and Henry's law. Non – ideal systems. Azeotropes: HCl-H₂O, ethanol – water systems. Fractional distillation. Partially miscible liquids – phenol – water, trimethyl amine – water system, Nicotine - water

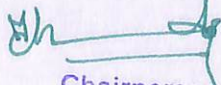
Lower and upper consolute temperature. Effect of impurity on consolute temperature. Immiscible liquids and steam distillation.

Symmetry of molecules (3 Hrs)

Symmetry operations and symmetry elements in molecules. Definition of Axis of symmetry types of C_n, Plane of symmetry (σ_h , σ_v , σ_d) Center of symmetry and improper rotational axis of symmetry (S_n). Explanation with examples.

Stereochemistry of Carbon Compounds (6 Hrs)

Conformations with respect to ethane, butane and cyclohexane. Molecular representation: Wedge Formula, Newmann, Sawhorse and Fischer representations. Optical isomerism: optical activity, optical rotation and specific rotation, Concept of chirality. Examples: Glyceraldehyde, Lactic acid, Alanine. Molecules with similar chiral carbons (Tartaric acid), Enantiomers and Meso compounds. Molecules with dissimilar chiral carbons (2,3 – Dibromopentane). Diastereomerism. Configuration: Relative (D and L) and Absolute configuration, CIP Rules: R/S Racemic mixture racemization and resolution techniques (chemical method only) Geometrical isomerism with reference to alkenes and cycloalkanes: cis – trans and E/Z configuration.


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ST.FRANCIS DEGREE COLLEGE FOR WOMEN BEGUMPET
HYDERABAD-500016
(AN AUTONOMOUS COLLEGE OF OSMANIA UNIVERSITY)
DEPARTMENT OF NUTRITION

DSC-II	NUTRITION FOR THE FAMILY SEMESTER- II	60 HRS
Module 1 - Balanced Diet and Meal Planning Module 2 - Pregnancy, Lactation and Infancy Module 3 - Preschoolers, School Going Children Module 4 - Adolescence and Geriatrics		

- The syllabus contains four Modules. Paper should give equal weightage to all Modules.
Four long questions- One question per module with internal choice

SEMESTER-I

NUTRITION FOR THE FAMILY

1. Course Description

Programme: B.Sc
Course Code: U24/NUT/DSC/102
Course Type: DSC -2
No. of credits: 4

Max. Hours: 60
Hours per week: 4
Max. Marks: 100

2. Course Objectives

- To gain knowledge about the nutritional requirements and the basics of meal planning for various physiological age groups
- To apply the principles and nutrition guidelines in meal planning in different stages of life cycle.

3. Course Outcomes

On completion of the course the student will be able to:

- CO 1: Understand the importance of a balanced diet, meal planning and nutritional requirements of adults.
CO 2: Understand and remember the different physiological changes during pregnancy, lactation and infancy.
CO 3: Understand the effect of nutrition on the growth and development of children
CO 4: Apply the guidelines and modifications of the diet during adolescence and old age period.

4. Course Content

MODULE I: BALANCED DIET AND MEAL PLANNING (15 Hrs)

- 1.1. Balanced Diet:** Definition, Benefits, Nutrient requirement, recommended dietary allowances (RDA), Basic four food groups, Food Exchange List, Food pyramid.
- 1.2. Meal planning:** Definition, Principles in meal planning, Points to be considered in planning a diet. Steps involved in planning a diet. Low Cost balanced diet.
- 1.3. Nutritional requirements of Adults:** RDA, Reference Man and Reference Woman. Nutritional and food requirements for an Adult man and woman according to Occupation.

MODULE II: PREGNANCY, LACTATION AND INFANCY (15 Hrs)

- 2.1. Pregnancy:** Physiological stages during pregnancy, Preconception nutrition status, Nutritional requirements, Dietary Guidelines, General Dietary problems, and complications during pregnancy.
- 2.2. Lactation:** Physiology of lactation, Nutritional requirements during lactation, Dietary guidelines.
- 2.3. Infancy:** Growth and development during infancy, Nutritional requirements during infancy, advantages of Human milk over bottle feeding, Artificial Feeding, Weaning foods

MODULE III: PRESCHOOLERS, SCHOOL GOING CHILDREN (15 Hrs)

- 3.1. Pre- School Children:** Milestones and growth chart, Nutritional requirements, factors to be considered while planning diets for preschoolers, dietary guidelines, Problems faced during feeding a preschooler.
- 3.2. School going children:** Nutritional requirements, food requirements, Packed lunch, factors to be considered while planning a packed lunch for school going children, Problems in feeding school going children.
- 3.3. School Lunch Programmes:** Objectives of feeding programmes. Intervention programmes in India, Mid - day meal programme, special nutrition programme, ICDS etc.

MODULE IV: ADOLESCENCE AND GERIATRICS (15 Hrs)

- 4.1 Adolescence:** Role of hormones on growth and development, sequence of developmental changes, nutritional requirements during adolescence, challenges in adolescence, weight control, skipping meals, anorexia, fast foods.
- 4.2 Geriatrics:** Physiological changes during ageing Changes in organ function with ageing, nutritional requirements and dietary modifications. Dietary guidelines.
- 4.3 Common diseases affecting geriatric:** Osteoporosis, obesity, constipation, malnutrition

5. References

1. Dietetics (sixth edition) B. Srilakshmi, New Age publishers.
2. Human Nutrition and Diet, M. Swaminathan, fifth edition, 2008.
3. Nutrition and dietetics, Shubhangini A. Joshi, 2012, (8th reprint).
4. Fundamentals of foods and nutrition, Sumathi R. Mudambi, 4* edition, 1990.
5. Human Nutrition second edition B. Srilakshmi, New Age publishers.
6. Nutrition - A life cycle approach - Ravinder Chadha, Orient Blackswan pvt.ltd.

6. Syllabus Focus

a) Relevance to Local, Regional, National and Global Development Needs

Local/Regional/National /Global Development Needs	Relevance
National needs	The course teaches about nutritional needs and emphasizes on planning a balanced diet for different age groups.

b) Components on Skill Development/Entrepreneurship Development/ Employability

SD/ED/EMP	Syllabus Content	Description of Activity
SD	1,2,3,4	Students are taught to plan balanced diet for different age groups (infancy - geriatrics)

7. Pedagogy

S. No	Student Centric Methods Adopted	Type/ Description of Activity
1.	Presentation , Assignments	Participative Learning
2.	Quiz	Experiential Learning
3.	Group Discussion, Seminar	Participative Learning

8. Course Assessment Plan

a) Weightage of Marks in Continuous Internal Assessments and End Semester Examination

CO	Continuous Internal Assessments CIA-50%	End Semester Examination-50%
CO1	CIA-1	Written Exam
CO2	CIA-1	
CO3	CIA-2 Breakfast foods	
CO4	CIA-2 Quiz/ cross word/MCQ	

SECTION A - INTERNAL CHOICE**4Q X 10M = 40 M**

Question Number	Question	Question	CO	BTL (Blooms Taxonomy Level)
1	Mod1	What is meal planning and its objectives?	CO 1	I
2	Mod 1	Write about the nutritional requirements for adults.	CO 1	II
3	Mod2	Explain the food and nutrient requirements during pregnancy.	CO 2	II
4	Mod 2	What makes breast milk an ideal food for infants?	CO 2	I
5	Mod 3	Describe the various School Lunch Programs.	CO 3	I
6	Mod3	Write in detail about the nutrient requirements and problems faced for a pre-school going child.	CO 3	II
7	Mod4	Explain the term Geriatric nutrition. What are the physiological changes occurring during aging?	CO 4	II
8	Mod 4	Elaborate the nutritional requirements during adolescence. Write a note on food choices and eating habits.	CO 4	VI

SECTION B - ANSWER ANY 4 OUT OF 5**4Q X 5 M= 20 M**(To compulsorily have **ONE** question from **each** module)

9	Module 2	Explain the physiology of Lactation	CO 2	II
10	Module 2	What is weaning foods	CO 2	I
11	Module 1	Discuss the Food Pyramid	CO 1	II
12	Module 1	What is Balanced Diet	CO 1	I
13	Module 3	Define Packed Lunch	CO 3	I
14	Module 4	Explain Osteoporosis	CO 4	I

**NUTRITION FOR THE FAMILY
PRACTICAL****Programme: B.Sc.****Course Code: U24/NUT/DSC/201/P****Course Type: DSC 2****No. of credits: 1****Max.Hours : 30****Hours per week: 3****Max.Marks: 50****Course Objectives:**

1. To know the qualitative analysis of carbohydrates, proteins and lipids.
2. Gain knowledge on functions, sources and effects of deficiency of nutrients.

Course Outcome:

On completion of the course the student will be able to:

- To apply the knowledge of planning diet charts of different age groups
- To understand the different age groups and their nutritional requirements

Practical Sessions

1. Adult man - sedentary
2. Adult woman – sedentary,
3. Pregnant woman- Moderate worker.
4. Lactating mother- Sedentary worker.
5. Preschool child
6. School going
7. Adolescent girl
8. Adolescent boy
9. Geriatrics
10. Planning and preparation of a weaning mixture.

NUTRITION FOR THE FAMILY
MODEL QUESTION PAPER
PRACTICAL




Course Code: U24/NUT/DSC/101/P
No. of credits: 1

Marks : 50
Time: 2 Hrs

Answer the following

Answer the following

1. Plan a day's diet for the given age group (A, B,C)20 M
- A.05M
- B.05M
- C.10M
- a. Introduction
- b. Write the RDA for the above mentioned age group
- c. Menu for the day
2. Calculate the nutritive value for the planned diet25M
- a) Calculations15 M
- b) Grand total table05 M
- c) Result05 M
- 3) Record05 M

Prepared by	Checked & Verified by	Approved by
 Ms Chhandasi Das Signature of the teaching faculty	 Ms. Tabitha Ramona Signature of HoD	 Dr. Uma Joseph Signature of Principal

St. FRANCIS COLLEGE FOR WOMEN, BEGUMPET, HYDERABAD-500016
(An Autonomous College Affiliated To Osmania University)
FACULTY OF SCIENCE- DEPARTMENT OF CHEMISTRY
PRACTICAL SYLLABUS CBCS-2024
SEMESTER -II
QUANTITATIVE ANALYSIS II
(Volumetric Analysis)

Program: B.Sc.
Course Code: U24/CHE/DSC/201/P
Course: DSC-2
No. of Credits: 1

Max. Hours: 20 Hrs
Max. Marks: 50
Hours per week: 2 Hrs

Course Objectives

- To develop analytical skills using the principles of quantitative analysis.

Course Outcomes

CO 1: Interpret and apply the principles of redox and complexometric titrations.

CO 2: Quantitative estimation of salts using gravimetric principles.

Volumetric Analysis

1. Estimation of Fe (II) ions by titrating it with $K_2Cr_2O_7$ using an internal indicator.
2. Estimation of Cu (II) ions using $Na_2S_2O_3$ with $K_2Cr_2O_7$ as primary standard.
3. Estimation of Iodine content in Iodized salt.
4. Estimation Nickel by back titration using $MgSO_4$.
5. Estimation of Zinc using EDTA.
6. Estimation of calcium or magnesium ions in milk.
7. Estimation of hardness of water.

References:

1. Vogel's Qualitative Inorganic Analysis, *Svehla, G.* Pearson Education, 2012.
2. J. Vogel's Quantitative Chemical Analysis, *Mendham*, Pearson, 2009.


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6. Syllabus Focus

a. Relevance to Local, Regional, National and Global Development Needs

Local /Regional/National /Global Development Needs	Relevance
local	In manufacturing processes, local energy production and storage techniques, production of solvents and other materials, drug formulation and dosage calculations.
Regional	Preparation of fertilizers impacting agriculture, metal refining process, food and pharmaceutical applications, useful for analytical techniques.
National	Optimizing the use of P - block elements, focus on electrochemical principles, preparation of polymers, phase transformations of various systems.
Global	Electronic industry, advancements in electrochemical technologies, sustainable industrial processes impacting the environment, formulation of solutions.

b. Components on Skill Development/Entrepreneurship Development/ Employability

SD/ED/EMP	Syllabus Content	Description of Activity
SD	Module No. 1	Encourage them to compare and discuss trends in reactivity and chemical behaviour
EMP	Module No. 2	Engage students in electrochemical cell design and analysis
ED	Module No. 3	Invite professionals or researchers who can share their insights from organic chemistry.
EMP	Module No. 4	Divide students into groups and share some real time applications.



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7. Pedagogy

S. No.	Student Centric Methods Adopted	Type / Description of Activity
1.	Experiential Learning	Field trips
2.	Participative Learning	Presentations
3.	Problem solving	Case studies

8. Course Assessment Plan

a. Weightage of Marks in Continuous Internal Assessments and End Semester Examination

CO	Continuous Internal Assessments CIA - 40%	End Semester Examination-60%
CO1	CIA 1 written exam	Written Exam
CO2	CIA 2 (Quiz/Assignment/3D model making)	
CO3	CIA 1 written exam	
CO4	CIA 2 (Crossword/Problem solving/Assignment)	


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b. Model Question Paper - End Semester Exam

St. FRANCIS COLLEGE FOR WOMEN, BEGUMPET, HYDERABAD-500016

(An Autonomous College Affiliated to Osmania University)

Faculty of Science – Department of Chemistry

MODEL PAPER

B. Sc. I YEAR SEMESTER -II

TIME: 2 hrs

Course Code: U24/CHE/DSC/201

Max. Marks: 60 M

SECTION –A (Essay Questions)

I. Answer the following

4X10M=40 Marks

1. a). Classify the oxides based on the oxygen content. (CO 1) L1 4M
b). What are interhalogen compounds? Explain the structure of the AX₅ type of molecule. (CO 1) L1 6M

OR

2. a). Discuss the structure of Diborane. (CO 1) L2 5M
b). What are silicones? Classify them. (CO 1) L1 5M
3. a). Describe the Hittorf method for the determination of transport number (CO 2) L2 5M
b) Explain Debye –Huckle's theory and the role of inter-ionic effect. (CO 2) L2 5M

OR

4. a) Explain the working and construction of S.H.E. (CO 2) L2 5M
b) State Kohlrausch law of independent migration of ions and list out its applications. (CO 2) L2 5M
5. a) Give the mechanism of Nitration on Benzene. (CO3) L1 5M
b) Write a note on directive influence of methyl group. (CO3) L1 5M

OR

6. Explain S_N1 and S_N2 reactions with their mechanism, stereochemistry, and energy profile diagram. (CO 3) L2 10M
7. a) What are azeotropes? Explain ethanol-water system. (CO 4) L2 5M
b) State and explain Raoult's law with its limitations. (CO 4) L2 5M

OR

8. a) Justify that the chair form of cyclohexane is the most stable conformation. (CO4) L4 6M
b) Define proper axis of symmetry. Illustrate with 2 examples. (CO4) L1 4M

SECTION – B (Short Answer Questions)

II. Answer any four

4x5=20 Marks

9. Discuss the structure of XeO₃. (CO1) L1
10. Calculate the EMF of Cd, Cd²⁺//Cu²⁺, Cu E⁰ (Cu²⁺, Cu) = 0.34V.
E⁰(Cd²⁺, Cd) = -0.488 V. (CO 2) L5
11. How can you interpret aromatic character in a molecule? (CO3) L5
12. State and explain Henry's Law and its limitations. (CO 4) L2
13. Write a note on Williamson synthesis. (CO3) L1
14. What are carbides and give their classification. (CO1) L1



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b. Model Question Paper - End Semester Exam

St. FRANCIS COLLEGE FOR WOMEN, BEGUMPET, HYDERABAD-500016

(An Autonomous College Affiliated to Osmania University)

Faculty of Science – Department of Chemistry

B. Sc. I YEAR SEMESTER -II

Time: 2 Hrs

Max. Marks: 60M

Course Code: U24/CHE/DSC/201

Credits: 4

SECTION A - INTERNAL CHOICE			4 Q X 10 M = 40M	
Question Number	Question		CO	BTL
1	Module 1	a) Classify the oxides based on the oxygen content. 4M b) What are interhalogen compounds? Explain the structure of AX ₃ type of molecule. 6M OR	CO 1	(Level II)
2	Module 1	a) Discuss the structure of Diborane. 5M b) What are silicones? Classify them 5M	CO 1	(Level II)
3	Module 2	a) Describe the Hittorf method for the determination of transport number 5M b) Explain Debye –Huckle's theory and explain the role of inter-ionic effect. 5M OR	CO 2	(Level II)
4	Module 2	a) Explain the working and construction of S.H.E. 5M b) State Kohlrausch law of independent migration of ions and list out application 5M	CO 2	(Level II)
5	Module 3	a) Give the mechanism of Nitration on Benzene. 5M b) Write a note on directive influence of methyl group. 5M OR	CO 3	(Level I)
6	Module 3	Explain the hydrolysis of primary and tertiary alkyl halides with mechanism, stereochemistry and energy profile Diagram. 10M	CO 3	(Level II)

7	Module 4	a) What are azeotropes? Explain ethanol-water system. 5M b) State and explain Raoult's law with its limitations. 5M OR	CO 4	(Level II)
8	Module 4	a) Justify that the chair form of cyclohexane is the most stable conformation. 6M b) Define proper axis of symmetry. Illustrate with 2 examples. 4M	CO 4	(Level IV)
SECTION B – (Short answer questions)				
ANSWER ANY 4 OUT OF 6			4 X 5M = 20 M	
9	Module 1	Discuss the structure of XeO_3	CO 1	(Level I)
10	Module 2	Calculate the EMF of $\text{Cd}, \text{Cd}^{2+}/\text{Cu}^{+2}, \text{Cu}$ $E^0(\text{Cu}^{+2}, \text{Cu}) = 0.34\text{V}$. $E^0(\text{Cd}^{2+}, \text{Cd}) = -0.488\text{V}$.	CO 2	(Level V)
11	Module 3	How can you interpret aromatic character in a molecule?	CO 3	(Level V)
12	Module 4	State and explain Henry's Law and its limitations.	CO 4	(Level II)
13	Module 3	Write a note on Williamson synthesis.	CO 3	(Level I)
14	Module 1	What are carbides and give their classification.	CO 1	(Level I)


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